ABSTRACT

Nine Zn(II) complexes of valproic acid and nitrogen-donor ligands, formulating **(1)**, $[Zn(valp)_2, 9-dmphen]$ (2), $[Zn(valp)_21,10-phen]$ as $[Zn_2(valp)_4]$ (3), $[Zn(valp)_2, 2-bipy]$ (4), $[Zn(valp)_24,4-bipy]_n$ **(5)**, $[Zn_2(valp)_4(quin)_2]$ **(6)**, $[Zn(valp)_2(2-ampy)_2]$ (7), $[Zn(valp)_2(2-ampic)_2]$ (8) and $[Zn(valp)_2(2-picam)]$ **(9)** were synthesized and characterized using IR, ¹H NMR, ¹³C{¹H} NMR and UV-Vis spectrometry. The crystal structures of the complexes 2, 6 and 7 were determined using single-crystal X-ray diffraction.

The in-vitro antibacterial activity of the prepared complexes was investigated against Gram-positive (M. luteus, S. aureus and B. subtilis) and Gram-negative (E. coli, K. pneumoniae and P. mirabilis) bacteria using agar diffusion method. All prepared complexes except complexes 7 and 8 exhibit antibacterial activity different Gram-positive and Gram-negative against bacteria. The effect of complexation on the antibacterial activity of the parent ligands investigated for complexes 1, 2, 3 and 6. The complexes 1 and 6 showed higher antibacterial activity than their parent valproate ligand. Complex 2 showed higher antibacterial activity against Gram-negative bacteria than 2,9-dmphen ligand. For Gram-positive bacteria the activity of 2 and 2,9-dmphen are similar. The antibacterial activity of 1,10-phen ligand against both Gram-positive and Gramnegative bacteria was decreased upon complexation with zinc valproate.